P. P. SAVANI UNIVERSITY

First Semester of M.Sc. Examination

February-2022
SSCH7010-Advances in Inorganic Chemistry-I
07.2.2022,Monday Time: 12:00 PM to 2:30PM Maximum Marks: 60

insu	uctions:	
1. T	he question paper comprises of two sections.	
2. N	lake suitable assumptions and draw neat figures wherever required.	
3. U	se of scientific calculator is allowed.	
	Section-I	
Q.1	Very Short Questions (Attempt any five)	[10]
1.1	What is Nucleophillic Substitution in VESPR theory.	
1.2	Write Advantages of LFT.	
1.3	Give the Comparison of CH ₄ , NH ₃ , H ₂ O and H ₃ O ⁺ .	
1.4	Give the Full form of LCAO.	
1.5	Give the full form of VESPR theory?	
1.6		1
1.0	Give the BENT RULE statement.	
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Q.2	Write Short Notes (Attempt any two)	[06]
2.1	Application of Gillespie Laws	
2.2	Formation of σ-molecular Orbital	
2.3	Square Planar Complexes	
Q.3	Detail questions (Attempt any two)	[14]
3.1	Comparison of PF ₅ , SF ₄ , ClF ₃ and [ICl ₂ -]	
3.2	dπ-pπBonding in Phosphorous Group Elements	
3.3	Explain Linear Combination of Atomic Orbitals (LCAO)	
	Section-II	
Q.1	Very Short Questions (Attempt any five)	[10]
1.1	What is group description of Redox reaction	[10]
1.2	Write the Born-Oppenheir approximation law.	
1.3		
	Write Thermodynamic Stability.	
1.4	Write all parameter in the given equation in $\Delta G = \Delta H - T \Delta S$.	
1.5	Write about Basicity of the ligands.	
1.6	What is Steric factor?	
Q.2	Write Short Notes (Attempt any two)	[06]
2.1	Aquation	
2.2	Mole Ratio Method	
2.3	Factors Affecting the Liability of Complex.	
Q.3	Detail questions (Attempt any two)	[14]
3.1	Thermodynamic Importance of Stability Constants	[]
3.2	Explain Spectrophotometric (a) Job's Method (b) Raghav Rao's Method	
33	Evalois MADCHE THEODY	